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Polar \u0026 Non-Polar Molecules: Crash Course Chemistry #23**12. The Shapes of Molecules: VSEPR Theory Polar Bonds, Polarity and Intermolecular Forces** **CHEMISTRY DK014 - TOPIC 7.2 (Part 1) - VSEPR Theory and 5 Basic Molecular Geometry** [Vsepr And Imf Homework](#)
VSEPR and IMF Homework Problems Lewis Structure Practice #1 Draw the Lewis structure for each of the following molecules: 1. Br 2 2. Methane 3. H2S 4. Nitrogen trichloride 5. Phosphorus pentabromide 6. Boron trihydride 7. Carbon disulfide Lewis Structure Practice #2 Draw the Lewis structure for each of the following molecules 1. Arsenic triiodide 2.

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